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Key indicators

Single-crystal X-ray study T = 294 K Mean σ (C–C) = 0.003 Å R factor = 0.044 wR factor = 0.115 Data-to-parameter ratio = 14.2

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

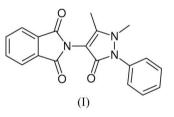
2-(1,5-Dimethyl-3-oxo-2-phenyl-2,3-dihydro-1*H*-pyrazol-4-yl)isoindoline-1,3-dione

In the title compound, $C_{19}H_{15}N_3O_3$, the phthalimide group, dihydropyrazole ring and phenyl ring are all essentially planar. The dihedral angles between the phthalimide group and dihydropyrazole ring, and the phthalimide group and phenyl ring are 57.0 (2) and 46.3 (3)°, respectively.

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Comment

Phthalimides and N-substituted phthalimides are an important class of compounds because of their interesting biological activities (Lima et al., 2002). In the title phthalimide derivative, (I) (Fig. 1), the phthalimide group is essentially planar to within 0.015(3) Å, and its geometry is closely comparable to in the related compounds that 2-(4-hydroxyphenyl)isoindoline-1,3-dione (Liang & Li, 2006a) and 2-ethylisoindoline-1,3-dione (Liang Li, 2006b). The & dihydropyrazole ring is planar to within 0.032 (3) Å and the phenyl ring is planar to within 0.009 (4) Å. The dihedral angles between the phthalimide group and dihydropyrazole ring, and the phthalimide group and phenyl ring are 57.0(2) and 46.3 $(3)^{\circ}$, respectively.



Experimental

A mixture of phthalimide (0.1 mol) and 4-amino-1,2-dihydro-2,3dimethyl-1-phenylpyrazol-5-one (0.1 mol) in acetic acid (90 ml) was refluxed for 4 h. After cooling, filtration and drying, the title compound was obtained. A sample of (I) (10 mg) was dissolved in acetone (13 ml) and the solution kept at room temperature for 5 d. Natural evaporation gave light-yellow single crystals suitable for X-ray analysis.

Crystal data $C_{19}H_{15}N_3O_3$ $M_r = 333.34$ Monoclinic, $P2_1/c$ a = 13.960 (3) Å b = 15.948 (3) Å c = 7.3536 (14) Å $\beta = 102.358$ (3)° V = 1599.2 (5) Å³

Z = 4 D_x = 1.385 Mg m⁻³ Mo K α radiation μ = 0.10 mm⁻¹ T = 294 (2) K Block, light yellow 0.22 × 0.18 × 0.12 mm

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Data collection

Bruker SMART CCD diffractometer φ and ω scans Absorption correction: multi-scan (*SADABS*; Bruker, 1997) $T_{\min} = 0.979, T_{\max} = 0.989$

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.044$ $wR(F^2) = 0.115$ S = 1.053257 reflections 229 parameters H-atom parameters constrained 8874 measured reflections 3257 independent reflections 2109 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.048$ $\theta_{\text{max}} = 26.4^{\circ}$

 $w = 1/[\sigma^{2}(F_{o}^{2}) + (0.0466P)^{2} + 0.256P]$ where $P = (F_{o}^{2} + 2F_{c}^{2})/3$ $(\Delta/\sigma)_{max} = 0.001$ $\Delta\rho_{max} = 0.21 \text{ e } \text{\AA}^{-3}$ $\Delta\rho_{min} = -0.17 \text{ e } \text{\AA}^{-3}$ Extinction correction: *SHELXL97* Extinction coefficient: 0.0184 (19)

H atoms were visible in difference Fourier maps, but were placed in idealized positions and refined using a riding model, with C-H = 0.93 Å and $U_{iso}(H) = 1.2U_{eq}(\text{aromatic C})$ or C-H = 0.96 Å and $U_{iso}(H) = 1.5U_{eq}(\text{methyl C})$. The methyl groups were allowed to rotate about their local threefold axes.

Data collection: *SMART* (Bruker, 1997); cell refinement: *SAINT* (Bruker, 1997); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1997); software used to prepare material for publication: *SHELXTL*.

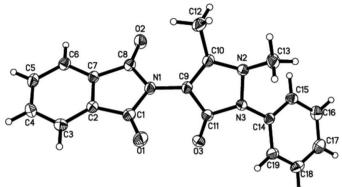


Figure 1

The molecular structure of (I), showing displacement ellipsoids drawn at the 30% probability level for non-H atoms.

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References

- Bruker (1997). SADABS, SMART, SAINT and SHELXTL. Bruker AXS Inc., Madison, Wisconsin, USA.
- Liang, Z.-P. & Li, J. (2006a). Acta Cryst. E62, 04436-04437.
- Liang, Z.-P. & Li, J. (2006b). Acta Cryst. E62, 05439-05440.
- Lima, L. M., Castro, P., Machado, A. L., Frage, C. A. M., Lugniur, C., Moraes, V. L. G. & Barreiro, E. (2002). J. Biol. Org. Med. Chem. 10, 3067–3073.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.